

**PHYSICAL CHEMISTRY III
2016-2017**

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| Bachelor Degree: | Chemistry | 702G |
| Course title: | Physical Chemistry III | 522 |
| Year/Semester: | Second Semester | ECTS Credits: 6 |

DEPARTMENT

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|---------------|------------------|----------------|---------------------|--------------|-------|
| Chemistry | | | | | |
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ENGLISH-FRIENDLY FACULTY

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CONTENTS

UNIT 1.- Surface tension and interfaces: Interfaces Thermodynamics, surface tension, adsorption processes, interfaces characterization techniques

UNIT 2.- Adsorption isotherms: gas-solid isotherms

UNIT 3.- Transport properties: Gas kinetics theory, transport properties in gas and solutions: diffusion, viscosity, electric, and thermal conductivities; ionic transport.

UNIT 4.- Electrical interfaces: Basic concepts, electrical interface thermodynamics, electrocapilarity, double layer models.

UNIT 5: Electrokinetics phenomena: Electroosmosis, electrophoresis. flux potential, sedimentation potential, electrodic reaction kinetics.

UNIT 6.- Macromolecules: Polimerization kinetics, characterization of polymers.

UNIT 7.- Colloids: Characterization of colloidal dispersions, colloidal stability, associate colloids.

Laboratory sessions include electrochemistry transport properties, isotherms, and associate colloids.

Computer laboratory sessions include electrical interfaces and electrokinetics phenomena simulations

REFERENCES

| Title |
|---|
| Atkins' Physical Chemistry. |
| Basic principles of colloid Science. |
| Interfacial Science: an introduction. |
| Modern electrochemistry. |
| Foundations of colloid Science. |
| Introduction to modern colloid Science. |
| Physical Chemistry for the life sciences. |
| Student's solutions manual to accompany Atkins' physical chemistry. |



The colloidal domain: where Physics, Chemistry, Biology, and Technology meet.

EVALUATION SYSTEM

Written exam – 70%

Oral presentation of laboratory session – 10%

Laboratory reports – 16%

Attitude evaluation in laboratory – 4%